

New Stereoselective Routes from Alcohols to Secondary Alkyl Bromides with Retention of Configuration

By MIREILLE SEVRIN and ALAIN KRIEF*†

(Department of Chemistry, Facultés Universitaires N.D. de la Paix, 61, rue de Bruxelles, Namur, Belgium)

Summary Secondary alcohols can be converted into the corresponding bromides with high stereoselective retention of configuration; this two-step process proceeds by a double inversion involving the intermediate selenides.

We report here a new method for converting secondary alcohols (A) into the corresponding alkyl bromides (C)^{1,2}

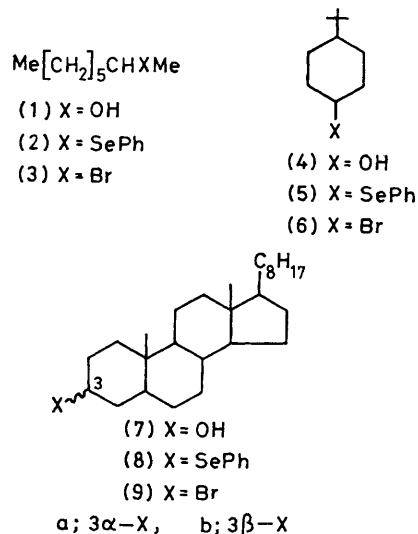
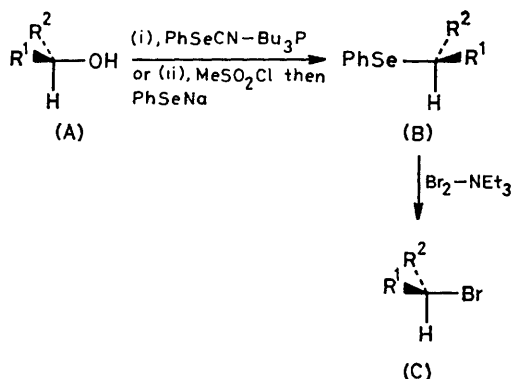


TABLE.

Starting alcohol ^a	Selenide	Yield/% (method)	$[\alpha]_D^{20}$ ^b (Stereochem. purity/%)	Bromide	Yield /%	$[\alpha]_D^{20}$ ^c (Stereochem. purity/%)
(-)-(R)-(1)	(+)-(S)-(2)	74 (i)	+30.95° (96.8)	(-)-(R)-(3)	90	-42.52° (95.5)
(-)-(R)-(1)	(+)-(S)-(2)	70 (ii)	+29.89° (93.5)	(-)-(R)-(3)	88	-41.32° (92.8)
(+)-(S)-(1)	(-)-(R)-(2)	79 (ii)	-29.80° (93.2)	(+)-(S)-(3)	89	+41.20° (92.52)
<i>cis</i> -(4)	<i>trans</i> -(5)	43 (i)	— (>95)	<i>cis</i> -(6)	95	— (>95)
<i>trans</i> -(4)	<i>cis</i> -(5)	71 (i)	— (>95)	<i>trans</i> -(6)	25	— (>95)
(7a)	(8b)	76 (ii)	— (>95)	(9a)	93	— (>95)
(7b)	(8a)	47 (i)	— (>95)	(9b)	22	— (>95)
		57 (ii)				

^a Purchased from Aldrich. The octanols used had the following $[\alpha]_D^{20}$ values: (+)-(S)-octan-2-ol 7.52° (optical purity 92.3%, ref. 2), (-)-(R)-octan-2-ol -7.00° (optical purity 85.8%). Optically pure octanols have $[\alpha]_D^{20} \pm 8.15^\circ$ (refs. 1 and 2). ^b Corrected for optically pure starting material in the case of the octan-2-ols. ^c Overall value for the whole process, corrected for optically pure starting materials in the case of the octan-2-ols.

† On leave of absence at: I.C.I., Plant Protection Division, Jealott's Hill, Berkshire.

with high (>92%) stereoselective retention of configuration at the substituted carbon atom. This unusual reaction[‡] takes advantage of the easy synthesis of phenyl selenides (B) from alcohols [PhSeCN-Bu₃P-tetrahydrofuran(THF), 20 °C, 72–100 h, method (i),³ or MeSO₂Cl-ether, 0 °C, followed by substitution by PhSeNa in THF–hexamethylphosphoric triamide, 20 °C, method (ii)] and their further transformation into the corresponding bromides⁴ (Br₂-NEt₃-CH₂Cl₂, 20 °C); both stages were found to proceed with high stereoselectivity and inversion of configuration at the substituted carbon atom.

The overall yields were usually good (ca. 60%; Table) except for the equatorial alcohols *trans*-(**4**) and (**7b**); in these

cases there was a competing elimination reaction during the selenide-alkyl bromide reaction.

The stereochemical purity of the bromides was ascertained by comparison with samples prepared unambiguously by known methods ([α]_D for the octyl bromides,^{2a,b} ¹H n.m.r. spectroscopy and g.c.² for the cyclohexyl bromides,^{5a-c} and ¹H n.m.r. spectroscopy for the cholesteryl bromides^{5d,e}).

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[‡] Alkyl chlorides, however, can be prepared from alcohols and thionyl chloride with varying degrees of inversion or retention of configuration depending on the reaction conditions (ref. 1).

¹ For a synthesis of optically active alkyl halides, see H. R. Hudson, *Synthesis*, 1969, 112.

² (a) D. Landini, S. Quici, and F. Rolla, *Synthesis*, 1975, 430; (b) J. San Filippo and L. J. Romano, *J. Org. Chem.*, 1975, **40**, 1514; (c) R. O. Hutchins, D. Masilamani, and C. A. Maryanoff, *ibid.*, 1976, **41**, 1071.

³ P. A. Grieco, S. Gilman, and M. Nishizawa, *J. Org. Chem.*, 1976, **41**, 1485.

⁴ M. Sevrin, W. Dumont, L. Hevesi, and A. Krief, *Tetrahedron Lett.*, 1976, 2647.

⁵ (a) E. L. Eliel and R. G. Haber, *J. Org. Chem.*, 1959, **24**, 143; (b) E. L. Eliel, *Chem. Ind. (London)*, 1959, 568; (c) E. L. Eliel and R. J. L. Martin, *J. Am. Chem. Soc.*, 1968, **90**, 689; (d) A. K. Bose and B. Lal, *Tetrahedron Lett.*, 1973, 3937; (e) G. Roberts, C. W. Shoppee, and R. J. Stephenson, *J. Chem. Soc.*, 1954, 2705.